COMPLETE LISTING OF CLAIMS, INCORPORATING AMENDMENTS IN RESPONSE TO OFFICE ACTION DATED May 4, 2006 FOR SERIAL NO.10/664,950

We claim:

- 1. (Currently amended) A process for removing acid anhydrides precursors and other acid moieties from flue gases, the process comprising:
- a) cooling the flue gases containing $\underline{CO_2}$, acid anhydride precursors and other acid moieties;
 - b) removing particulate matter from the flue gases;
- c) oxidizing the acid anhydrides precursors to oxidized anhydrides, wherein the acid anhydrides precursors are selected from the group consisting of SO₂ and NO_x, and the acid moieties are selected from the group consisting of HCI, HF and hydrogen halides;
- d) simultaneously converting the oxidized anhydrides <u>precursors</u> and the other acid moieties to ammonia compounds/ salts <u>by reaction with aqua ammonia compounds</u>; and
- e) <u>subsequently converting CO₂ to an ammonia salt by reaction with aqua</u> ammonia compounds; and
 - <u>f)</u> collecting the ammonia compounds/ salts.
- 2. (Original) The process as recited in claim 1 wherein gaseous anhydrides are oxidized to higher gaseous acid anhydrides.
- 3. (Previously Amended) The process as recited in claim 1 wherein the step of simultaneously converting oxidized anhydrides and the other acid moieties includes reacting the oxidized anhydrides with ammonia-containing compounds.

- 4. (Original) The process as recited in claim 3 wherein the ammonia-containing compounds are water-soluble compounds selected from the group consisting of aqueous ammonia, ammonium hydroxide, ammonium carbonate, ammonium carbamate, and combinations thereof.
- 5. (Original) The process as recited in claim 3 wherein the ammonia-containing compounds are regenerated by the thermal decomposition of ammonium bicarbonate (NH₄HCO₃) to carbon dioxide (CO₂), ammonia solution (NH₄OH), ammonium carbonate ((NH₄)₂CO₃), and combinations thereof.
- 6. (Original) The process as recited in claim 1 wherein the process is carried out at temperatures of from about 15°C to 50°C.
- 7. (Original) The process as recited in claim 5 wherein the regeneration is carried out at a temperature from about 35°C to 80°C.
- 8. (Previously amended) The process as recited in claim 5 wherein ammonia and ammonia-containing compounds generated from the decomposition of ammonium bicarbonate are recycled for use in the conversion step.
- 9. (Original) The process as recited in claim 1 wherein the other acid moieties are hydrogen halides.

- 10. (Withdrawn) A device for the direct removal of acid anhydrides and other acid moieties from a gas stream, the device comprising:
 - a) a means for cooling the gas stream;
 - b) a means for eliminating particulate matter from the gas stream after cooling;
- c) a means for oxidizing the anhydrides present in the gas stream after the removal of particulate matter;
 - d) a means for converting the oxidized anhydrides to salts;
 - e) a means for regenerating the converting means; and
 - f) a means for isolating the salts from the gas stream.
- 11. (Withdrawn) The device as recited in claim 10 wherein the means for converting the oxidized moieties is a first scrubber containing a neutralizing agent.
- 12. (Withdrawn) The device as recited in claim 11 wherein the neutralizing agent contains aqueous ammonia.
- 13. (Withdrawn) The device as recited in claim 10 wherein the means for regenerating the converting means comprises heat.
- 14. (Withdrawn) The device as recited in claim 10 wherein the temperature range of operation of the device is from about 15°C to 50°C.

- 15. (Withdrawn) The device as recited in claim 10 wherein the regeneration is carried out at a temperature from about 35° C to 80° C.
- 16. (Withdrawn) The device as recited in claim 10 wherein the other acid moieties are hydrogen halides.

| 17. (Withdrawn) The device as recited in claim 10 wherein a product produced |
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| by the means for regeneration is recycled back to the means for converting. |
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| 18. (Cancel) |
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| 19. (Cancel) |
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| 20. (Previously presented) The process of claim 3 further comprising the |

regenerating the ammonia-containing compounds.